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Book : : Introduction to Chemical Engineering Thermodynamics, J. M. Smith, H. C. Van Ness, M. M. Abbott, and M. T. Swihart, 8th edition, McGraw-Hill, New York, 2018.

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Equation (6.66b) therefore becomes:  $GR = Z \cdot 1 \cdot \ln(Z \cdot q^* RT \cdot ?) \cdot Z$  For given T and P, Z is found by solution of Eq. (3.52) for a vapor phase or Eq. (3.56) for a liquid phase with  $? = ? = 0$ .

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