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Name: \_\_\_\_\_ Stoichiometry Sheet 6 Please  
answer the following questions about the  
combustion of acetylene:  $2\text{C}_2\text{H}_2 + 5\text{O}_2 \rightarrow$   
 $4\text{CO}_2 + 2\text{H}_2\text{O}$  1. How many moles of  
acetylene ( $\text{C}_2\text{H}_2$ ) are needed to  
completely react with 12 moles of  
oxygen? 2. How many grams of carbon  
dioxide are produced from the combustion  
of 1.3 moles of acetylene

Name: Stoichiometry Sheet 1 Mass / Mole

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