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2015, Before 5 p.m. Problem  
3: Bohr's model for the H  
atom still sees the electron  
"orbiting" on classical  
trajectories around the  
proton, it only restricts  
the possible radii.

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(B) From solution of  $V_0(R) =$   
0 34. In the electronic  
transition pictured below,  
the molecule starts in its  
ground vibrational level.

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The most probable vibrational quantum number of the excited state is (A) 1 (B) 2 (C) 3 (D) 4 (E) 5 (D) 5th level,  $v = 4$  35.

Which of the following is a true statement about the fluorescence of a

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Prerequisites: Math 220 and  
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**Chem 332: Physical Chemistry**

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The textbook for this course is the fifth edition of 'Quantum Chemistry' by Ira N. Levine. You can get used copies on Amazon at a reasonable price. You can probably find copies around the chemistry building if you are resourceful. Exams, Quizzes, Papers & Other Major Graded Work There are two in-class exams and a final.

Quantum Chemistry: Through Problems & Solutions  
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